

Información del Plan Docente

Academic Year	2017/18
Faculty / School	100 - Facultad de Ciencias
Degree	540 - Master's in Industrial Chemistry
ECTS	6.0
Year	1
Semester	Second semester
Subject Type	Compulsory
Module	---

1.General information**1.1.Introduction****1.2.Recommendations to take this course****1.3.Context and importance of this course in the degree****1.4.Activities and key dates****2.Learning goals****2.1.Learning goals****2.2.Importance of learning goals****3.Aims of the course and competences****3.1.Aims of the course****3.2.Competences****4.Assessment (1st and 2nd call)****4.1.Assessment tasks (description of tasks, marking system and assessment criteria)****5.Methodology, learning tasks, syllabus and resources****5.1.Methodological overview**

The learning process designed for this course combines theoretical concepts with assessment activities and the solving of practical cases close to reality. All these activities will be combined in a strategic way with the student's autonomous work to achieve the best academic output.

In particular, the following activities will be carried out: theory sessions, problem sets and case study, laboratory sessions, assessment tasks, study and reports.

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- The theory sessions will be expositive, and in them the fundamental aspects of the course will be developed through the use of the necessary audiovisual/bibliographic media.
- In the problem sets and case study sessions, the student's active participation will be encouraged and valued.
- During the laboratory sessions, special attention will be paid to the correct management of the instruments, the techniques for obtaining experimental data and their validation and subsequent use in the calculation of magnitudes of interest. Unsolved problems and lab sessions will be carried out in small groups for an individualized assistance of the student.

5.2.Learning tasks

The course includes 6.0 ECTS organized according to:

- **Theory Sessions (1.4 ECTS): 36 h.** See the course syllabus
- **Problem Sets and Case Study (0.5 ECTS): 12 h.** Individual solving of advanced electrochemical problems related to electrosynthesis, electrodeposits, corrosion, electrochemical energy converters, reactions and photochemical processes and environmental photochemistry.
- **Laboratory Sessions (0.3 ECTS): 8 h.** Lab sessions show the experimental techniques of working with electrochemical and photochemical reactors.
- **Assessment Tasks (0.2 ECTS): 4 h.** Classroom presentation of assignments and reports.
- **Study and Reports (3.6 ECTS): 90 h.** Autonomous work of the student to carry out the assignments, problem sets, preparation of lab sessions and reports.

5.3.Syllabus

The course will address the following topics:

Topic 1. Fundamentals of the Electrochemical Synthesis.

- Basic concepts in electrochemical synthesis. Thermodynamic aspects. The electrical double layer. The kinetics of the electrode processes. Transport phenomena. Electrochemical adsorption. Interpreting the components of the cell potential.

Topic 2. Electrochemical Reactors. Components and Operation.

- Basic components in an electrochemical reactor. Physicochemical properties of common solvents and solvent selection. The supporting electrolyte. Types of electrodes and the selection criteria. Main materials for anodes and cathodes. Counter electrodes. Recent developments on electrode materials.
- Mass transport in the electrochemical reactors. Current and potential distributions. Geometry of reactors. Electric connections. Hydraulic distribution. Heat removal. Types of reactors and selection criteria.

Topic 3. Industrial Traditional Applications of the Electrochemical Synthesis.

- The chlor-alkali industry. Fluorine production. Hall-Heroult process of aluminium. Sodium chlorate manufacturing.

Topic 4. Modern Industrial Applications of the Electrochemical Synthesis.

- Production of adiponitrile, a key ingredient for nylon 6,6. Anisaldehyde production (perfume industry). Production of L-cysteine. Maltol production. Electrosynthesis in the pharmaceutical industry.

Topic 5. Electrochemical Synthesis and Environment.

- Intrinsic advantages of the electrochemical synthesis for the environment. Green electrochemical processes. Electrochemical decontamination.

Topic 6. Electroplating and Other Related Electrochemical Applications.

- General Aspects of electrodeposition. The various steps in the process of electrodeposition of metals. Crystal structure of the metallic electrodeposits. Parameters affecting the process. Additives (brighteners, levelers,

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surfactants and ductilizers).

- Industrial electroplating of Zn, Sn, latón, Cu, Ni y Cr. Modern electro-coatings with gold and silver.
- Other electrochemical applications: electroforming, electrowinning and electrorefining of metals; electromachining and electropolishing.

Topic 7. Corrosion and Corrosion Prevention.

- The effects and economic impact of corrosion. Types of corrosion.
- Electrochemical and no electrochemical methods for the corrosion rate measurement.
- The corrosion prevention; actions concerning the metal and actions concerning the corrosive medium (inorganic and organic coatings, pH control, corrosion inhibitors, cathodic and anodic protection). Practical applications in industry.

Topic 8. Electrochemical Energy Conversion and Storage.

- Types of electrochemical converters: primary cells, rechargeable batteries and fuel cells. Electrochemical behaviour of primary cells. Leclanché dry cell, alkaline batteries, silver oxide batteries (Zn-Ag₂O). Characteristics of commonly used rechargeable batteries (lead-acid, Ni-Cd, Ni-metal hydride, Li-ion batteries). New developments in fuel cell technology.

Topic 9. Fundamentals of Industrial Photochemistry.

- Photochemical activation of molecules and atoms. Primary and secondary photochemical processes. Mechanisms of deactivation. Quantum yield. The rate of the photochemical reactions. Effect of solvent. Sensitization. Experimental techniques.
- Photochemical synthesis. Photochemical reactors. Photochemistry in industrial synthesis.

Topic 10. Other Applications and Effects of Light. Photochemistry and Environment.

- Photoinitiators in the polymerization processes. Photochemical depolymerisation of plastics. Sunscreen and sun protection. Photochromism.
- Environmental photochemistry. Photochemical origin of some chemicals in the troposphere. Kinetics and mechanism of the main tropospheric reactions. Estimating the rate constants and half-lives for the reactions of organic molecules with HO· radicals in the troposphere. Photochemical processes for water treatment.

5.4.Course planning and calendar

Further information concerning the timetable, classroom, assessment dates and other details regarding this course, will be provided on the first day of class or please refer to the "Facultad de Ciencias" website <https://ciencias.unizar.es/node/7073>

5.5.Bibliography and recommended resources

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- Hamann, Carl H.. Electrochemistry / Carl H. Hamann, Andrew Hamnett and Wolf Vielstich . - 2nd completely rev. and updated ed. Weinheim : Wiley-VCH, cop. 2007
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- Coeuret, F.. Introducción a la ingeniería electroquímica / F. Coeuret ; edición coordinada por J. Costa López Barcelona [etc.] : Reverté, D.L. 1992
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 - Montalti, M.. Handbook of Photochemistry. 3ª CRC Press. 2013